Correction to the Structure of the Photo-isomers of Pyridazine 1,2-Dioxides; X-Ray Crystal Structure Analysis

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Summary The 1,4,6,7-dioxadiazocin structure (3), which was originally assigned to the photo-isomers of pyridazine 1,2-dioxide (1), is incorrect; chemical data and X-ray analysis show the correct structure to be 3a,6a-dihydro-isoxazolo[5,4-d]isoxazole (2).

IN a previous paper¹ we assigned the 1,4,6,7-dioxadiazocin structures (3) to the photo-isomers of pyridazine 1,2-dioxides (1). However, we have now found that the ¹H-n.m.r. data of the reduced products (4) from the photo-isomers and of their acetates (5) are inconsistent with the corresponding structures derived from the assumed structure (3).

The photo-isomer (2a) obtained from (1a) was treated with NaBH₄ in EtOH-tetrahydrofuran(THF) (1:5) at 0-5 °C to give a dihydro compound [(4a); oil, yield 10%]. Its n.m.r. spectrum showed geminally coupled signals at δ 3·18 (1H, dd, J 13·6 and 3·6 Hz, 3-H) and 3·56 (1H, br.d, J 13·6 Hz, 3-H'), an NH signal at 5·15 (1H, br.s), a signal at 5·35 (1H, dd, J 6·4 and 3·6 Hz, 3a-H) coupled with that of 3-H (J 3·6 Hz) and 6a-H at 5·67(1H, br.d, J 6·4 Hz), and a signal at 7·22 (1H, br.s, 6-H). The n.m.r. spectrum of the



acetate [(5a); m.p. 104—105 °C, ν (C=O) 1670 cm⁻¹ (KBr), yield 95%] obtained from (4a) with Ac₂O exhibited signals at δ 2·10 (3H, s, COMe), 3·38 (1H, dd, *J* 13·0 and 4·0 Hz, 3-H), 4·68 (1H, br.d, *J* 13·0 Hz, 3-H'), 5·38 (1H, dd, *J* 7·0 and 4·0 Hz, 3a-H), 5·74 (1H, br.d, *J* 7·0 Hz, 6a-H), and 7·35 (1H, br.s, 6-H).



FIGURE. Molecular structure of (2a).

These data indicate the presence of an ABX spin system in both (4a) and (5a), suggesting that a -CH-CH-CH(H')unit is present in each structure, and are quite inconsistent with the expected reduction product from (3) and its acetate for the eight-membered ring structure.

Thus, the photo-isomer must have the 3a,6a-dihydroisoxazolo[5,4-d]isoxazole structure (2a) which agrees with its ¹H- and ¹³C-n.m.r. spectra† and the ¹H-n.m.r. spectra of its reduction product. Analogous results were obtained when $R^1 = Me.\ddagger$

In addition, the structure of (2a) has been confirmed by single-crystal X-ray diffraction. Crystal data: $C_4H_4N_2O_2$, colourless needles from Pr_2^1O , space group $P2_1/n$, $a = 8\cdot 24$, $b = 11\cdot 17$, $c = 5\cdot 35$ Å, $\beta = 102\cdot 75^\circ$, Z = 4. Intensity data were collected using Cu- K_{α} radiation on a Phillips automatic four-circle diffractometer. 565 independent reflections with $I > 3\sigma(I)$ and $\theta < 60^\circ$ were used in the solution. The structure was solved by the direct method,² and refined by block diagonal least-squares to a final *R*-index of 0.06.§

As a mechanism for the formation of (2), we postulate that fission of the N-N bond of (1) by irradiation gives the bis-iminoxyl radical (6),³ followed by ring closure via (7) to give (2). Cyclization of the dinitroso compound (8), which has not yet been isolated, is also a possibility.

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[†] The original assignment of the structure (3) was principally based on the ¹H- and ¹³C-n.m.r. spectra as follows: the ¹H-n.m.r. spectrum of the photo-isomer of (1a) showed two sharp singlets at δ 6.05 (2H) and 7.38 (2H), assigned to two methine protons; its ¹⁵C-n.m.r. spectrum showed two signals (δ from Me₄Si) at 88.38 (=CH-) and 143.86 (=CH-). The photo-isomer of (1b) showed two signals in the ¹H-n.m.r. spectrum at δ 2.10 (6H, s) and 5.75 (2H, s); the ¹³C-n.m.r. spectrum showed three signals at 10.56 (Me), 89.98 (=CH-), and 153.74 (=CH-).

 \ddagger ¹H-n.m.r. data of (4b): δ 1·30 (3H, d, J 7·0 Hz, 3-Me), 2·08 (3H, s, 6-Me), 3·40 (1H, m, 3-H), 5·07 (1H, dd, J 7·0 and 4·0 Hz, 3a-H), and 5·49 (1H, d, J 7·0 Hz, 6a-H). ¹H-n.m.r. data of (5b): δ 1·42 (3H, d, J 7·0 Hz, 3-Me), 2·10 (3H, s, N-Ac or 6-Me), 2·12 (3H, s, N-Ac or 6-Me), 4·30 (1H, m, 3-H), 5·21 (1H, dd, J 7·0 and 6·0 Hz, 3a-H), and 5·48 (1H, d, J 7·0 Hz, 6a-H). The high-resolution mass spectrum of (2b) showed that the base peak of m/e 82 had been incorrectly assigned to C₄H₆N₂ in the previous paper (see ref. 1). The observed mass 82·0285 is assigned to C₄H₄NO (Calc. 82·0291, M^+ – MeCNOH).

§ The atomic co-ordinates for this work are available on request from the Director of the Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW. Any request should be accompanied by the full literature citation for this communication.

¹ H. Arai, A. Ohsawa, K. Saiki, and H. Igeta, J.C.S. Chem. Comm., 1977, 133.

²G. Germain, P. Main, and M. M. Woolfson, Acta Cryst., 1971, A27, 368.

³ J. L. Brokenshire, J. R. Roberts, and K. U. Ingold, J. Amer. Chem. Soc., 1972, 94, 7074; R. N. Butler, F. L. Scott, and T. A. F. O'mahony, Chem. Rev., 1973, 73, 93.

⁴ G. Maier and M. Wiessler, Tetrahedron Letters, 1969, 4987.